

HOMOCYSTEINE THIOLACTONE AS PRECURSOR OF METHIONINE AMIDE : APPLICATION
TO THE MODIFICATION OF PEPTIDES OF THE TACHYKININ FAMILY

G. Chassaing, S. Lavielle, S. Julien and A. Marquet

Laboratoire de Chimie Organique Biologique, U.A. C.N.R.S.

040493, Université P. & M. Curie, 4 Place Jussieu, Paris Cedex 05

Abstract

Two pathways for the substitution of a tert-butyl group from (S-tert-butyl)-homocysteine by an alkyl residue are described. Pathway B via homocysteine thiolactone, can be used to modify any peptide containing a C-terminal methionine amide.

The neuropeptide Substance P, (SP), is a member of the tachykinin family, a group of peptides which share, among other common features, a C-terminal methionine amide (1). Recently, two other tachykinins, NKA and NKB, (2-5) have been found in the central nervous system and the functions of these peptides have to be investigated. This prompted us to develop a general methodology for transforming the methionine residue into a large variety of analogues modified on sulfur or/and on the terminal carboxamide, in order to study structure-activity relationships or to obtain labelled ligands.

Substance P (SP) : Arg-Pro-Lys-Pro-Gln-Gln-Phe-Phe-Gly-Leu-Met-NH₂
Neurokinin α (NKA) : His-Lys-Thr-Asp-Ser-Phe-Val-Gly-Leu-Met-NH₂
Neurokinin β (NKB) : Asp-Met-His-Asp-Phe-Phe-Val-Gly-Leu-Met-NH₂

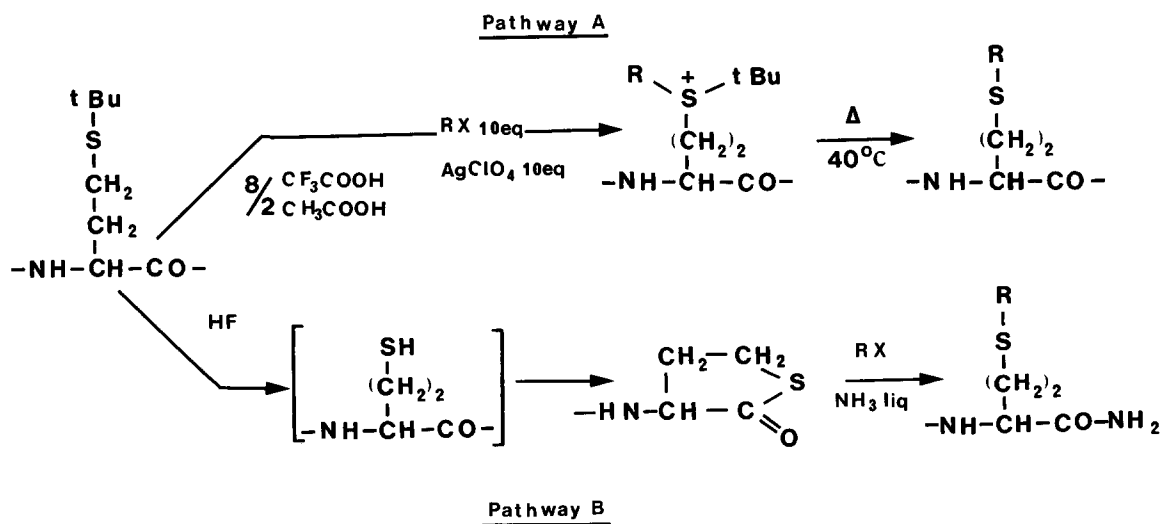
We described recently the two-step conversion of methionine into (S-tert-butyl)-homocysteine (6). In the present report, we propose two pathways for the substitution of

the tert-butyl group by an alkyl residue. In this study conducted to test the method, the alkylating agent was either methyl iodide or trideuteriomethyl p-toluene sulfonate.

Pathway A takes advantage of the easy thermal decomposition of tert-butyl sulfonium salts. It relies however on the possibility of obtaining the dissymmetric sulfonium salts.

For instance, the alkylation of (S-tert-butyl)-homocysteine by ICH_3 (50 equiv) in H_2O , at room temperature for 2 hours, led to the dimethyl sulfonium salt. The dissymmetric sulfonium could be obtained using 10 equiv. of the $\text{ICH}_3/\text{AgClO}_4$ complex at lower temperature (5°C) in $\text{CF}_3\text{COOH}/\text{CH}_3\text{COOH}$ (2:8). After two hours, where the excess of ICH_3 has been consumed further addition of H_2O and warming for 2 hours at 40°C yielded methionine with an excellent yield (90%).

The same conditions applied to {S-tert-butyl-Hcy¹¹} SP induced the precipitation of the peptide, and by increasing the concentration of CF_3COOH , the precipitation of AgClO_4 . This can be overcome by using AgBF_4 , more soluble in trifluoroacetic acid. However, in less than 5 min at 5°C in CF_3COOH we detected only the dimethyl sulfonium salt coming from the alkylation of the intermediate sulfide by either ICH_3 or the dissymmetric sulfonium



Scheme 1

salt. (7). As it has been observed that the electrophilicity of these salts is very depen-

dent on the various counter-ions and solvents (8), we checked a range of conditions. In the best that we found {CF₃COOH/CH₃SO₃H (10:1) with 10 equiv. of ICH₃/AgClO₄} we obtained a 4:1 ratio of symmetric and dissymmetric sulfoniums. Thus, we had to use another route.

Pathway B : The intermediate thiol {Hcy¹¹}SP generated by HF cleavage of { S-tert-butyl-Hcy¹¹}SP from the methylbenzhydrylamine polymer (MBHA) cannot be isolated and cyclises into the corresponding thiolactone peptide. We have previously attributed the thiol structure to the intermediate on the basis of amino-acid analysis and of its alkylation in liquid ammonia by methyl chloride (6). In fact, the isolated {Hcy¹¹}SP thiolactone intermediate was opened in liquid ammonia yielding the C-terminal amide and the thiol function thus generated was alkylated in situ by CH₃Cl.

The thiolactones of methionine, SP, NKA and NKB have been obtained from the corresponding StBu precursors. The peptides have been synthesized by solid-phase, the first residue being always N- α -Boc-S(tert-butyl)-homocysteine, linked to a MBHA polymer. After cleavage by HF and usual work-up, the crude peptides have been purified by classical techniques using either acidic or neutral eluants. The thiolactone function has been characterized by 250 MHz ¹H-NMR (two multiplets 3.25-3.17 ppm and 2.09 ppm respectively for the γ -CH₂ and β -CH₂ of the thiolactone ring) and by U.V. (the UV difference spectra between the thiolactone and the corresponding C-terminal methionine amide peptides present the same maximum at 238 nm (ϵ =3600)). The structure was confirmed by FAB Mass Spectrometry: {Hcy¹¹}SP thiolactone (MH⁺, m/z 1316) ; {Hcy¹⁰}NKA thiolactone (MH⁺, m/z 1102) ; {Hcy¹⁰}NKB thiolactone (MH⁺, m/z 1181).

The two methods we have described are complementary. Pathway A, when possible, is preferable, because of simpler experimental conditions. Pathway B is more general. The homocysteine thiolactone at the C-terminal position of a peptide is a stable precursor of methionine amide residue. It can be used for labelling a C-terminal methionine amide with a radioactive group and for the obtention of S modified analogs. It can also be opened by different nucleophiles to obtain C-terminal modified analogs.

Acknowledgment : we are indebted to Mr D. Godin for carrying out the FAB Mass Spectral Analysis.

References

1. Skrabaneck, P. and Powell, D., Annual Research Reviews (1983) Substance P, Vol 13, Eden Press Inc. (Horrobin, D.F. Ed.).
2. Kimura, S., Okada, M., Sugita, Y., Kanazawa, I. and Munekata, E., (1983). Proc., Japan Acad., 59, Ser. B, 101.
3. Kangawa, K., Minamino, N., Fukuda, A. and Matsuo, H., (1983) Biochem. Biophys. Res. Commun. 114, 533.
4. In accordance to the recommendation of the committee from the IUPHAR Satellite Symposium on Substance P, August 6th, 1984, Neurokinin α , identical to Substance K, and Neurokinin β , identical to Neurokinin K, have been named NKA and NKB respectively.
5. Nawa, H., Hirose, T., Takashina, H., Inayarra, S. and Nakanishi, S., (1983) Nature 306, 32.
6. Chassaing, G., Lavielle, S. and Marquet, A., (1983) J. Org. Chem., 48, 1760.
7. The symmetric and dissymmetric sulfonium were observed by HPLC ; C18 μ -Bondapak, 0.25 M triethylammonium phosphate buffer pH 3.0, 18% CH_3CN .
8. Badet, B., Julia, M. and Ramirez-Munoz, M., (1980) Synthesis 11, 926.

(Received in France 22 October 1984)